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# BIOETHANOL PRODUCTION FROM SUGARCANE BAGASSE PRETREATED BY TRICHODERMA VIRIDE

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Preservation of fossil fuels are currently depleting with the massive exploitation of fuels. In this condition, breakthroughs are necessary to produce alternative fuels. One of the breakthroughs is bioethanol. It is a renewable energy which is more effective than gasoline inasmuch it can increase combustion efficiency and reduce exhaust emissions. In this work, a bioethanol process was done by using sugarcane bagasse waste material which has a lot of lignin and cellulose content. The content was converted into bioethanol by utilizing a strong base to degrade lignin and T. Viride as cellulose-producing and S. cereviseae yeast as a sugar converter to bioethanol. This present research aims to find the best formula of bioethanol production based on sugarcane bagasse with variations in cellulose hydrolysis temperature, shaking speed, and fermentation time by using an integrated shaker machine fuzzy-logic control of temperature and humidity. This research employed a complete randomized design experimental research (CRD) to test temperature, speed, and time modification by using a shaker machine. The independent variables were: (1) temperature, (2) shaking speed, and (3) fermentation time. The dependent variables measured were reducing sugar and bioethanol levels. The results showed that the best formula for producing bioethanol levels was at a treatment temperature of 45 °C and a speed of 140 rpm with fermentation time of 48 hours which resulted in a bioethanol level of 2.75%.

Key words: bioethanol, sugarcane bagasse, reducing sugar, pretreatment, microbial

# INTRODUCTION

Almost all economic activities require a supply of fuel as energy and most of the fuel used is fossil. If exploited on a large scale, it results in depletion of fossil fuel reserves and rising prices. According to data from the Ministry of Energy and Mineral Resources in 2018, Indonesia only has oil reserves of around 3.3 billion barrels, assuming a day of crude oil production of 800 thousand barrels. As a result, in the next 11-12 years oil reserves in Indonesia will be vanishing.

One attempt to reduce dependence on fossil fuels is to employ renewable energy that can replace fossil fuels such as bioethanol, biosolar, biohydrogen, and others. Bioethanol is a renewable fuel that is quite promising and more effective than gasoline inasmuch it can increase combustion efficiency and reduce exhaust emissions [1]. Bioethanol can be made by using materials containing lignocelluloses which, when reduced, will diminish sugars in the forms of glucose, mannose, xylose, galactose and arabinose, and can be converted to bioethanol [2,3].

Production of plant-based bioethanol which can still

be consumed is ineffective and can cause food problems; therefore, we need more realistic bioethanol raw materials such as the rest of the agricultural and plantation production such as rice straw, corncobs, and bagasse which is one of the sufficient available materials nowadays. Indonesia has a very high potential since it has 450000 ha of sugarcane planted with sugar production of 2.53 million tons in 2015 [4]. Meanwhile, sugarcane mill waste reaches 14 - 40% by weight sugarcane as a sugar ingredient. This volume is potential enough to be used as a bioethanol material [5].

Sugarcane bagasse has around 50% cellulose. This content is higher than other ingredients such as corncobs which only have 40% cellulose content [6]. This means that the reducing sugar content obtained from saccharification process can be fermented to produce higher bioethanol, assuming 80 liters of bioethanol is produced from 1 ton of sugarcane. If sugarcane productivity is 80 tons per hectare, each hectare of sugarcane land produces 6400 liters of bioethanol. Bioethanol from sugarcane can substitute 20% of gasoline needs [5,7]. Increasing levels of bioethanol is imperative in order to obtain increased efficiency of bioethanol yield. Some techniques used to obtain efficient ethanol yield can be completed with preliminary stages of physical, physical-chemical, and biological processes. Physical engineering is usually carried out by mechanical breakdown in order to obtain small materials so as to facilitate the process of breaking down cellulose in materials. Meanwhile, chemical techniques are usually with acid hydrolysis, alkaline hydrolysis, and many more. In term of biological treatment, many microorganisms appear such as molds from species such as Polyporusbrumalis, and Lentinusedodes [8].



Research conducted by [9] investigated the xylanases produced by Trichoderma viride and Trichoderma piluliferum. Both the enzyme extracts produced were effective at increasing digestibility of sugarcane silage and corn silage. The enzymatic extracts proved promising for xylooligo-saccharides production and showed low cellulase activity, indicating potential for application in biobleaching processes of cellulose. Furthermore, research conducted by [10] found that solid culture experiments demonstrated Trichoderma viride and A. terreus as the most compatible strains with Leptosphaerulina sp. The use of fungal co-culture as inducers obtained superior results than previously used chemical inducers. These results revealed the potential of co-cultivation as an alternative for enzymatic induction and pollutants bioremediation. But, in their study [9, 10] not varying the treatment on shaker machines. According to [11] the use of shakers serves as a stirring so that the reagent substances can collide well. It is also contended that the higher the hydrolysis temperature, the better the saccharification process is [12].

Given the aforementioned complexities above, this present study seeks to find the best formula of bioethanol production based on sugarcane bagasse with variations in cellulose hydrolysis temperature, shaking speed, and fermentation time by using an integrated shaker machine that controls fuzzy-logic temperature and humidity.

# MATERIALS AND METHODS

# Materials

The object used in this research was bagasse in the form of waste in the process of making sugar from Mitra Agung Sugar Factory in Malang, East Java, Indonesia. Waste material of bagasse was dried by using an oven at 80  $^{\circ}$ C for 48 h. The dried material was then blended until it was crushed and sieved with a size of 40 mesh.

Trichoderma viride and Saccharomyces cereviseae cultures were bred in the Microbiology Laboratory of Universitas Negeri Malang, East Java, Indonesia. The culture preparations in this study included Trichoderma viride and Saccharomyces cereviseae on new medium so that they could grow optimally.

# Culture preparation of microbial

Trichoderma viride was cultured on PDA media and incubated at 28°C for 9 days. Furthermore, a starter 1 was made by adding 3 ose Trichoderma viride/ Saccharomyces cereviseae into 15 ml of the dissolved material. Starter 1 was placed in a shaker with a speed of 120 rpm for 24 h. Furthermore, a starter 2 was done by adding 3 ml of starter 1 to 27 ml of the material solution. Starter 2 was placed on a shaker with a speed of 120 rpm for 24 h. Furthermore, a starter 3 was done by adding starter 2 as much as 10% of the material to be used in the treatment then it was placed in a shaker machine with a speed of 120 rpm in 96 h for Trichoderma viride and 120 rpm in 48 h for Saccharomyces cereviseae.

# Chemical pretreatment

Chemical pretreatment was done to remove the lignin content in the ingredients, and at this stage, 50 g of bagasse powder was soaked in 1 liter of natrium chloride for 5 h. Besides, the powder was washed until the smell diminished. Furthermore, sterilization was performed by using an autoclave at 121 °C for 15 min.

# **Bioethanol production**

Bioethanol production in this research was carried out by saccharification and fermentation processes. Trichoderma viride and yeast cultures were done simultaneously through Saccarification and Fermentation (SSF) processes. The SSF method was done by giving Trichoderma viride and Saccharomyces cereviseae together. The shaking treatment was carried out by using an integrated Fuzzy Logic temperature and humidity shaker machine according to the study design. The culture technique was carried out by adding 5 ml each of the 3 Trichoderma viride and Saccharomyces cereviseae starter to 30 ml of treated sugarcane bagasse material and 10 ml of Andreoti media in a modified fermenter bottle. The pH is adjusted to 4 by adding 3M Hydrochloric acid Hydrolysis and fermentation were carried out at temperatures according to the study design of 25, 35, and 45 °C, with variations in speeds of 60, 100 and 140 rpm and lengths of 24, 48, and 72 h.

# Analytical

Measurement of the sugars levels was carried out by using Nelason method. Measurement of the ethanol levels was carried out by using a pocket refractometer.

#### **Research Design**

This study employed an experimental design to test the modification of temperature, speed and time of fermentation by using a completely randomized design (CRD) with twenty-seven treatments and two replications. The independent variables were: (1) temperature, (2) shaking speed, and (3) fermentation and the dependent variable measured was the level of reducing sugars and the level of bioethanol produced.

#### **RESULTS AND DISCUSSION**

#### Reducing sugar production

Reducing sugar was converted into bioethanol by using Saccharomyces cerevisiae. The more the reducing sugarproduced, the higher the level of bioethanol is [13]. To construe the trend of the enzymatic activity of Trichoderma viride in producing reducing sugars, saccharification was carried out at 25 °C with a speed of 100 rpm based on saccharification timing process. In terms of the acquisition of reducing sugar levels at a temperature of 25 °C with a speed of 100 rpm for 24 hours, it generates the lowest reducing sugar levels of 4162.83 µg, while



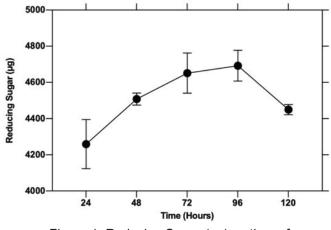


Figure 1: Reducing Sugar (µg) vs time of saccharification time

the highest reducing sugar levels of 4752.43  $\mu$ g were obtained at 96 h. It then decreased at the treatment time of 120 h to 4429.96  $\mu$ g. Data from the measurement of Reducing sugar content on bagasse using the Nelson method is presented in Figure 1.

Reducing sugar was produced through saccharification and simplified cellulose processes which was converted into simple sugars in the form of glucose [14]. Reducing sugar can be formed due to the enzymatic reaction produced by Trichoderma viride which produces the cellulase enzyme. Cellulase enzymes are a group of enzymes consisting of several enzymes working synergistically to break down cellulose. The enzyme group is the endoglucanase enzyme that randomly changes cellulose and forms a free end chain. The exoglunase enzyme degrades molecules that have been broken down by endoglucasease and removes the cellubiosa unit from the free end and the last enzyme,  $\beta$ -glucosidase, converts the cellubiose into a simpler sugar glucose [3,15].

This research was carried out through saccharification on bagasse delignified by using Trichoderma viride which was shaken by using a shaker machine with a speed of 100 rpm at a temperature of 25 °C within 24h long. It was obtained from the reducing sugar amounted to 4354.94 μg. During 48 h, 4531.35 μg was obtained. Meanwhile 4729.76 µg was gained within 72 h. The increase still continued at 96 h with 4752.43 µg. These data indicate that there was an increase in reducing sugar levels at each 24 h treatment period. The increase was due to an increase in cellulase enzyme activity; but after the 120 hour treatment, the reducing sugar showed a decrease. This occurred inasmuch the activity of the cellulase enzyme decreased with increasing hours. It is showed that the optimum time of the saccharification process with the help of molds was 48 h after which it would decrease its reducing sugar levels [10].

Other factors influencing the results of the reducing sugars are the amount of mold inoculants and the temperature of the saccharification process. Factors of mold inoculants will be directly related to the concentration of enzymes produced by these molds since the more molds used, the more enzymes produced. Thus, the saccharification process will be faster. The reaction rate can be increased according to the increase in the concentration of the enzyme used [14]. The temperature factor is also one of the factors determining the outcome of saccharification. It is contended that the optimal reaction rate was around 45-50 °C. On the other hand, excessive temperature rise will damage the enzymes used for the saccharification process inasmuch enzymes are composed of proteins that will be denatured at high temperatures [16].

#### **Bioethanol production**

Data from the measurement of bioethanol by using the simultaneous saccharification fermentation (SSF) method at temperature, speed and time of treatment are presented in Figure 2.

The results of this study indicate that the best bioethanol production formula made from bagasse has a bioethanol content of 2.75% at 45 °C, 140 rpm shaking speed, and 48 h fermentation time fermented by using an integrated shaker machine fuzzy-logic control temperature and humidity. Furthermore, lowest bioethanol levels are 0% at a temperature of 25 °C, a speed of 60 rpm, and 24 h fermentation time.

The fermentation process is a process of changing simple sugars produced from the saccharification process into bioethanol. The processes used were simultaneous saccharification and fermentation (SSF). The SSF method is considered better than other methods inasmuch it increases the rate of the fermentation process since there is integration between the processes of saccharification and fermentation simultaneously and because the sugar formed from the direct saccharification process can be used as a raw material in the fermentation process. Thus, the emergence of acetic acid inhibitors can be avoided [17].

This study also indicates that the highest bioethanol content of 2.75% is obtained from the treatment at a temperature of 45 °C with a shaking speed of 140 rpm shaker

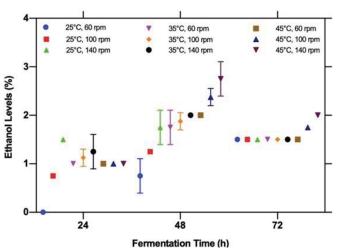
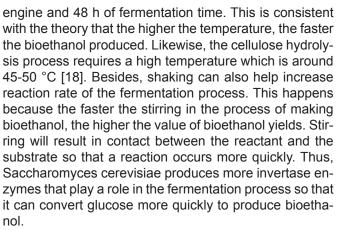


Figure 2: Effect of temperature, shaking speed an fermentation time on ethanol levels



This study, additionally, portrayed that the highest levels of bioethanol were obtained at the time of fermentation for 48 h and would decrease at 72 h. This happens because the best time for fermentation is 50 h [17]. The longer fermentation will reduce the rate of bioethanol fermentation since the nutrients present in the media decreases so that Saccharomyces cerevisiae will not be able to convert glucose into bioethanol. In addition, if fermentation is carried out longer, the alcohol content will actually decrease as alcohol will be converted into other compounds such as acetic acid and esters [19].

# CONCLUSION

Based on this study, the best formula to produce the highest levels of bioethanol of 2.75% was obtained at 45 °C with 48h fementation time by using an integrated shaker engine controlling temperature and humidity at a speed of 140 rpm. This study then encourages further research on the other microbial-aided bioethanol, the method of saccharification and fermentation in stages, and bioethanol made from other types of raw materials that can reduce environmental waste.

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